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SITU.DWPI,EPAB,JPAB,USPT,PGPB.	130002
SITUS.DWPI,EPAB,JPAB,USPT,PGPB.	3235
(1 SAME (IN ADJ SITU)).USPT,PGPB,JPAB,EPAB,DWPI.	0

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JPO Abstracts Database
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Derwent World Patents Index

Database: IBM Technical Disclosure Bulletins

11 same (in situ)

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DB Name	Query	Hit Count	Set Name
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USPT,PGPB,JPAB,EPAB,DWPI	11 same (fix\$ or cross link)	2	<u>L2</u>
USPT,PGPB,JPAB,EPAB,DWPI	glutaraldehyde bisulfite	110	<u>L1</u>

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Search Results - Record(s) 1 through 2 of 2 returned.**1. Document ID: US 5811529 A**

L2: Entry 1 of 2

File: USPT

Sep 22, 1998

US-PAT-NO: 5811529

DOCUMENT-IDENTIFIER: US 5811529 A

TITLE: Bis-pyridone compounds

DATE-ISSUED: September 22, 1998

INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Vines; Danette R.	Charlotte	NC		
Pedemonte; Ronald P.	Eppstein-Vockenhausen			DEX

US-CL-CURRENT: 534/560; 534/561**ABSTRACT:**

A fiber reactive dye having the formula ##STR1## wherein R.sub.1 and R.sub.3 are independently selected from the group consisting of; hydrogen; an alkyl group having from one to four carbon atoms; the foregoing alkyl group further substituted with one or more groups selected from hydroxy; amino; sulfo; halogen; an aryl radical; a heterocyclic radical or combination thereof; where for R.sub.3 said amino group maybe further substituted by a nitrogen heterocyclic fiber reactive group of the series: 1, 3, 5 mono or dichloro triazinyl; 1, 3, 5 mono or difluoro triazinyl; trichloropyrimidinyl; difluoropyrimidinyl; or monochlorodifluoro pyrimidinyl; where the nitrogen heterocycle may be further substituted by an: alkyl; or aryl amino group;

wherein B is selected from the group consisting of a C.sub.2 to C.sub.4 alkyl chain; a substituted aryl or a heterocyclic radical;

wherein X is selected from the group consisting of SO.sub.3 H; SO.sub.3 Na or a hydroxy group; and wherein Q.sub.1 and Q.sub.2 are independently selected from diazo components.

8 Claims, 0 Drawing figures Exemplary Claim Number: 1

L2: Entry 1 of 2

File: USPT

Sep 22, 1998

DOCUMENT-IDENTIFIER: US 5811529 A
TITLE: Bis-pyridone compounds

BSPR:

The structure of the coupling component of the present invention incorporates advantages over the prior art. For example, this coupling agent includes a bridge, whereby two solubilizing groups are incorporated in the dyestuff molecule by one condensation reaction. Not only does the bridge enhance solubility through the sulfonic acid moieties present, but it also provides a new way to incorporate two mono-reactive dyes producing one larger dye with improved levels of fixation. In the case of the aliphatic bridge, such as the glutaraldehyde bisulphite condensation product, the fixation properties should be further enhanced. The improved solubility should make these new dyes less prone to precipitate out of solution upon the introduction of salt and alkali. Further, the bridge links two reactive chromophoric systems together which should in theory provide greater tinctorial strength.

Full	Title	Citation	Front	Review	Classification	Date	Reference	Claims	Full	Draw Desc	Image
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2. Document ID: US 4266010 A

L2: Entry 2 of 2

File: USPT

May 5, 1981

US-PAT-NO: 4266010

DOCUMENT-IDENTIFIER: US 4266010 A

TITLE: Silver halide photographic light-sensitive material

DATE-ISSUED: May 5, 1981

INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Nagatomo; Shigeru	Minami-ashigara			JPX
Hori; Kiyotaka	Minami-ashigara			JPX

US-CL-CURRENT: 430/355; 430/539, 430/543, 430/642, 430/950, 430/961

ABSTRACT:

A silver halide photographic light-sensitive material containing at least one photographic layer containing acid-processed gelatin and a matting agent.

10 Claims, 0 Drawing figures Exemplary Claim Number: 1

L2: Entry 2 of 2

File: USPT

May 5, 1981

DOCUMENT-IDENTIFIER: US 4266010 A

TITLE: Silver halide photographic light-sensitive material

DETL:

Processing Step Development 25 seconds
Fixation 25 seconds Washing 20 seconds Developer Composition Sodium Sulfite 40 g
Hydroquinone 25 g Boric Acid 10 g 1-Phenyl-3-pyrazolidone 1.5 g Potassium
Hydroxide 30 g 5-Methylbenzotriazole 0.15 g Glutaraldehyde-bisulfite 15 g Acetic
Acid 12 g Potassium Bromide 5 g Water to make 1 l Fixing Solution Ammonium
Thiosulfate 174 g Sodium Sulfite (anhydrous) 20 g Sodium Tetraborate
(decahydrate) 20 g Acetic Acid 25 g Sulfuric Acid 5 g Aluminum Sulfate 7 g Water
to make 1 l

Full	Title	Citation	Front	Review	Classification	Date	Reference	Claims	PublC	Draw Desc	Image
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Term	Documents
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FIX.DWPI,EPAB,JPAB,USPT,PGPB.	449669
FIXA.DWPI,EPAB,JPAB,USPT,PGPB.	111
FIXAB.DWPI,EPAB,JPAB,USPT,PGPB.	1
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FIXABI.DWPI,EPAB,JPAB,USPT,PGPB.	4
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FIXABILIBY.DWPI,EPAB,JPAB,USPT,PGPB.	1
(L1 SAME (FIX\$ OR CROSS LINK))USPT,PGPB,JPAB,EPAB,DWPI.	2

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=> s glutaraldehyde bisulfite

L1 3 GLUTARALDEHYDE BISULFITE

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=> d 1-2 bib ab

L2 ANSWER 1 OF 2 BIOSIS COPYRIGHT 2001 BIOSIS
AN 2000:400550 BIOSIS
DN PREV200000400550
TI Disinfectant composition.
AU Synodis, Joseph; Wilensky, Stuart (1); Halecky, Alan
CS (1) Matawan, NJ USA
ASSIGNEE: Block Drug Company, Inc.
PI US 6034138 March 07, 2000
SO Official Gazette of the United States Patent and Trademark Office Patents,
(Mar. 7, 2000) Vol. 1232, No. 1, pp. No pagination. e-file.
ISSN: 0098-1133.
DT Patent
LA English
AB The present invention comprises a concentrated solid or semi-solid
disinfectant or sterilant composition for use in an aqueous disinfecting
or sterilizing solution, comprising an oxidant and a protected
glutaraldehyde such as a **glutaraldehyde bisulfite**
addition compound (GBS): ##STR1## or a glutaraldehyde dioxime compound
(GDO): ##STR2## The present invention further provides a method for
disinfecting or sterilizing a surface or apparatus comprising the steps of
mixing a concentrated solid or semi-solid glutaraldehyde sterilant
composition comprising an oxidizing compound and a protected sterilant
with water to form a solution and bringing the solution into contact with
the surface or apparatus.

L2 ANSWER 2 OF 2 MEDLINE DUPLICATE 1
AN 96190982 MEDLINE
DN 96190982 PubMed ID: 8604152
TI Inactivation of glutaraldehyde by reaction with sodium bisulfite.
AU Jordan S L; Russo M R; Blessing R L; Theis A B
CS Union Carbide Corporation, Bound Brook, NJ 08805, USA.
SO JOURNAL OF TOXICOLOGY AND ENVIRONMENTAL HEALTH, (1996 Feb 23) 47 (3)
299-309.
Journal code: KAA; 7513622. ISSN: 0098-4108.
CY United States
DT Journal; Article; (JOURNAL ARTICLE)
LA English
FS Priority Journals
EM 199605
ED Entered STN: 19960524
Last Updated on STN: 19960524
Entered Medline: 19960516
AB The microbiocidal activity of glutaraldehyde was inactivated by reaction
with sodium bisulfite via formation of a proposed **glutaraldehyde**
-bisulfite complex. High-performance liquid chromatography
(HPLC) analysis of 2% (0.2M) alkaline glutaraldehyde indicated complete
loss of glutaraldehyde at a 2.2:1 molar ratio of sodium bisulfite to
glutaraldehyde. Neither 1.7% (0.17 M) sodium bisulfite alone nor the
glutaraldehyde-bisulfite complex was microbiocidal when
tested against Escherichia coli, Pseudomonas aeruginosa, Enterobacter
aerogenes, and Polybac Polyseed BOD seed inoculum. Bacterial inhibition
tests indicated that the glutaraldehyde-sodium bisulfite complex had no
effect on the growth of sewage microorganisms at concentrations as high as
50-100 ppm (5×10^{-4} - 1×10^{-3} M), with an IC₅₀ of 230-440 ppm ($2.3 \times$
 10^{-3} - 4.4×10^{-3} M), based on glutaraldehyde concentration. A 28-close
bottle test showed a 5-d biodegradation of 48% and 51%, and a 15-d
biodegradation of 57% and 63% for 3:1 and 2.2:1 bisulfite to
glutaraldehyde molar ratios, respectively. Acute aquatic toxicity testing
with Daphnia magna demonstrated an LC₅₀ of 41-109 ppm (4.1×10^{-4} - $10.9 \times$
 10^{-4} M) and a no-observed-effect concentration (NOEC) of 16 ppm ($1.6 \times$
 10^{-4} M) for the proposed **glutaraldehyde-bisulfite**
complex (based on glutaraldehyde concentration), approximately 10-fold
higher than found for glutaraldehyde alone, indicating that the proposed

glutaraldehyde-bisulfite complex is less toxic to the environment than glutaraldehyde.

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E2	17	SZEKEPCZES J/AU
E3	1 -->	SZEKEPES/AU
E4	8	SZEKEPES A/AU
E5	12	SZEKEPES A V/AU
E6	1	SZEKEPES ANDRAS/AU
E7	115	SZEKEPES BARTHO J/AU
E8	21	SZEKEPES BARTHO JULIA/AU
E9	2	SZEKEPES C K/AU
E10	4	SZEKEPES CHARLES K/AU
E11	1	SZEKEPES D/AU
E12	30	SZEKEPES E/AU

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E13	3	FOURBIL Y/AU
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E15	4 -->	FOURCADE/AU
E16	102	FOURCADE A/AU
E17	5	FOURCADE ALAIN/AU
E18	1	FOURCADE AMAD EO M L/AU
E19	1	FOURCADE AMADEO M L/AU
E20	112	FOURCADE C/AU
E21	1	FOURCADE C J/AU
E22	9	FOURCADE CHRISTINE/AU
E23	16	FOURCADE D/AU
E24	22	FOURCADE E/AU

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)

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L3 0 L2 AND LOWRY

=> s l2 and lowy

L4 2 L2 AND LOWY

=> d 1-2 bib ab

L4 ANSWER 1 OF 2 MEDLINE

AN 95312436 MEDLINE

DN 95312436 PubMed ID: 7540753

TI Bone marrow one step fixation-decalcification in **Lowy** FMA
solution: an immunohistological and in situ hybridization study.

AU Gaulier A; **Fourcade C**; Szekeres G; Pulik M

CS Service d'Anatomie Cytologie Pathologiques, C. H. Victor Dupouy,
Argenteuil, France.

SO PATHOLOGY, RESEARCH AND PRACTICE, (1994 Dec) 190 (12) 1149-61.

Journal code: PBZ; 7806109. ISSN: 0344-0338.

CY GERMANY: Germany, Federal Republic of

DT Journal; Article; (JOURNAL ARTICLE)

LA English

FS Priority Journals

EM 199507

ED Entered STN: 19950807

Last Updated on STN: 19960129

Entered Medline: 19950727

AB The immunoreactivity of paraffin embedded bone marrow biopsies (BMB) was
studied following a one step 20-hour-fixation-decalcification in
Lowy formalin mercuric chlorid acid solution which permits
excellent histological stainings. Antibodies reactive with myeloid,
megakaryocytic, erythroid cells, T and B lymphocytes, mastocytes and
metastatic cells were compared. Nearly all antibodies working on paraffin
sections were demonstrated on **Lowy** FMA fixed BMB. Special care
was taken to define an optimal working dilution. Trypsinization was not
necessary. A slide microwave pre-treatment appeared essential before

-- testing CD20 L26, CD8, CD3, CD34, MB1 Kappa and Lambda antibodies. It was suitable for UCHL1, LN2, CD30 antibodies. The same fixative allowed an mRNA Kappa or Lambda in myeloma and EBER 1 EBV RNAs in HIV lymphoma visualization by in situ hybridization. The safety handling of the toxic mercuric chloride component is discussed.

L4 ANSWER 2 OF 2 BIOSIS COPYRIGHT 2001 BIOSIS
AN 1995:166744 BIOSIS
DN PREV199598181044
TI Bone marrow one step fixation-decalcification in **lowy** FMA
solution: An immunohistological and in situ hybridization study.
AU Gaulier, A. (1); **Fourcade, C.**; Szekeres, G.; Pulik, M.
CS (1) Serv. d'Anat. Cytol. Pathol., C.H. Victor Dupouy, 69 rue du Lt-Cl
Prudhon, 95107 Argenteuil Cedex France
SO Pathology Research and Practice, (1994) Vol. 190, No. 12, pp. 1149-1161.
ISSN: 0344-0338.
DT Article
LA English
AB The immunoreactivity of paraffin embedded bone marrow biopsies (BMB) was studied following a one step 20-hour-fixation-decalcification in **Lowy** formalin mercuric chloride acid solution which permits excellent histological stainings. Antibodies reactive with myeloid, megakaryocytic, erythroid cells, T and B lymphocytes, mastocytes and metastatic cells were compared. Nearly all antibodies working on paraffin sections were demonstrated on **Lowy** FMA fixed BMB. Special care was taken to define an optimal working dilution. Trypsinization was not necessary. A slide microwave pre-treatment appeared essential before testing CD20 L26, CD8, CD3, CD34, MB1 Kappa and Lambda antibodies. It was suitable for UCHL1, LN2, CD30 antibodies. The same fixative allowed an mRNA Kappa or Lambda in myeloma and EBER 1 EBV RNAs in HIV lymphoma visualization by in situ hybridization. The safety handling of the toxic mercuric chloride component is discussed.

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